

Synthesis, Structure, Magnetic and Electric Transport Properties of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$

PAWAN K. SHARMA¹ AND INDU B. SHARMA²

¹Govt. Degree College Khour, Jammu, 181203, India.

²ISCAS, Institute of Solid State and Material Science, Jammu University Campus, Jammu.
corresponding author email: drpksharma59@gmail.com

Abstract

A new phase with the composition $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$ has been synthesised by the standard ceramic method. X-ray diffraction studies show that the phase crystallizes in hexagonal unit cell ($a=17.452\text{\AA}$ and $c=6.463\text{\AA}$). The molar magnetic susceptibility measurements as a function of temperature suggest that the phase is diamagnetic and magnetic susceptibility is temperature independent. The electrical resistivity measurements as a function of temperature suggest that the phase is semi-conductor in nature in the temperature range 300K-500K and the conduction occurs via thermally activated mechanism. Anomaly at around 344 K, shows a variation in the energy of activation. The thermal analysis suggests that above 673 K there is mass loss of about 4% and at low temperature (496.9K), there is phase transformation which results polymerisation process.

Keywords: Mixed binary dichalcogenides, XRD, electrical resistivity, TG-DTA

Introduction:

Binary dichalcogenides of numerous elements with composition MX_2 and their mixed analogues $\text{M}_{1-x}\text{M}'_x\text{X}_2$ (M and M' are different transition elements; X=S, Se or Te) are known in the literature [1, 2]. Many dichalcogenides with reduced content of X are also known [3, 4]. It has been reported that structure and physical properties substantially vary with change in composition [1, 2, 3, 4]. It was thought interesting to prepare mixed chalcogenides with composition $\text{M}_{0.5}\text{M}'_{0.5}\text{X}_2$ study of their crystal structure & follow their physical properties as function of temperature.

In the present study, synthesis of a new phase with the composition $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_2$ has been reported. Its crystal structure has been determined from the powder X-ray diffraction data. Magnetic and electric transport properties have been studied in the temperature range 80K-300K and 300K-500K respectively. The phase has been analyzed for thermogravimetric analysis (TGA) and differential thermogravimetric analysis (DTGA).

Experiment

Synthesis Aldrich make Molybdenum (Mo) Silicon (Si) and Selenium (Se) elements (purity 99.9%) have been used for synthesis of the new phase. The constituent elements weighed corresponding to the stoichiometry $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$ were mixed and homogenized by grinding in cyclohexane. The dried and homogenized mixture, pressed into pellets in hydraulic press was placed in quartz tube and

evacuated to $\sim 10^{-5}$ Torr, vacuum sealed and was heat- treated at 1048K for 72 hours. The mixture during the heat treatment was subjected to a number of intermediate grindings, pelletizing and sealing undersame conditions for the completion of the reaction. The final product was pulverized to fine powder for further investigations [5, 6, 7].

Elemental Analysis

The phase was further analyzed by atomic absorption spectrophotometry, which is one of the most prevalent methods for the trace element analysis [8, 9, 10]. The results of chemical elemental analysis [11, 12] and the atomic absorption spectrophotometry are in good agreement. The data are given in Table 1.

Table 1: Analytical data of the phase ($\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$).
The theoretical value is given parenthesis. Analysis (%)

Phase	Mo	Si	Se
$\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$	21.57 (21.81)	6.26 (6.38)	69.64 (71.80)

X-ray Diffraction studies

Room temperature powder X-ray diffraction data of the product were recorded on a Stoe-powder diffraction system and a Philips diffractometer at a scanning speed of 1deg./minute in the 2θ range using $\text{CuK}\alpha$ and $\text{FeK}\alpha$ radiations [13, 14 and 15]. The X- ray diffraction data are given in the Table 2, while the X-ray pattern, intensity, versus 2θ is drawn in the figure1.

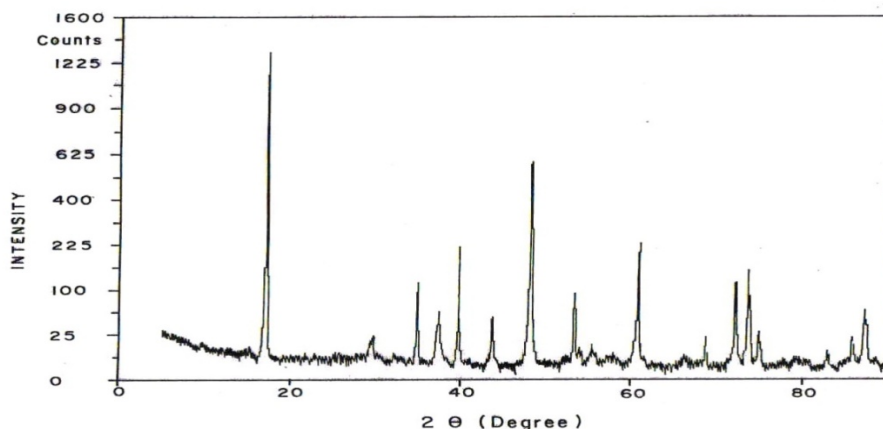


Figure 1: X-ray Diffraction pattern of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$

Magnetic Susceptibility Measurement

Magnetic susceptibility of the powdered phase was recorded in a Faraday balance provided with Polytronic Faraday-type electromagnet and a Mettler microbalance. Specially fabricated Dewar flask of the size which could be adjusted within pole gaps of electromagnet was used for keeping liquid nitrogen, which surrounded the phase crucible [16, 17]. The phase was held hanging in the inner tube of the Dewar

flask with a fine thread. Magnetic susceptibility in the temperature range 77K-300K could be measured by this arrangement.

Electrical Resistance study

Electrical Resistivity of thin pellet of the phase as a function of temperature in a continues flow of nitrogen was recorded by four probe method in a four probe cell, using Keithley programmable constant current supply source model 224 and nanovoltmeter model 181 for the purpose of current source and voltage measurement respectively [18,19]. The bottom surface of the pellet was kept non-conducting. The data of specific resistance (ρ) as a function of temperature are given in Table 4, while the $\log \rho$ versus $1/T$ data are plotted in the figure2.

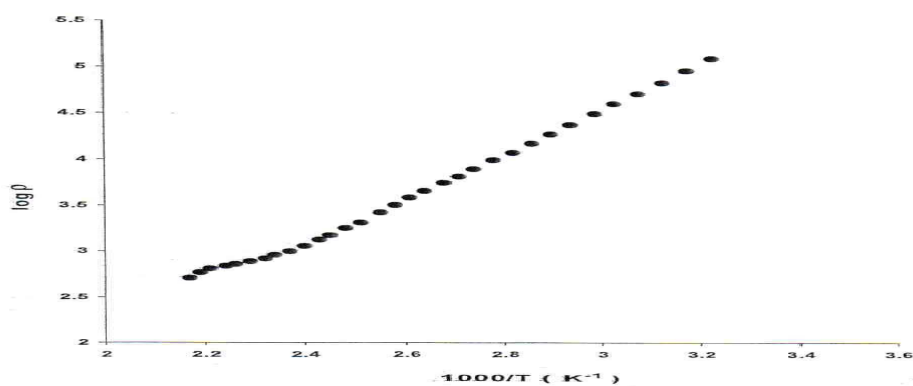


Figure 2: Log ρ versus $1/T$ plot of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$

Thermal Analysis

The phase has been thermally analyzed for thermogravimetric (TG) and differential thermal analysis (DTA) by Rigaku Thermal Analysis System 8150, provided with a microprocessor, in the temperature range 300K-873K at the heating rate of 10 deg./min in continuous flow of nitrogen [20, 21]. The TG and DTA plot is given in figure 3.

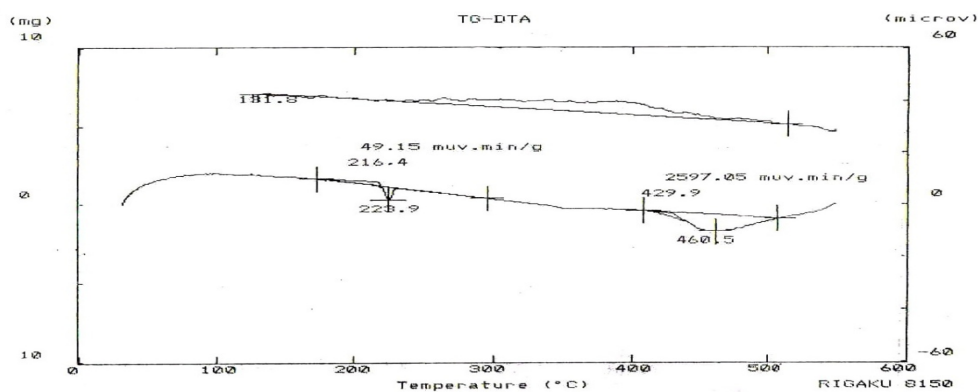


Figure 3: TG-DTA Curves of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$

Results and Discussion

Crystal Structure

The unit cell parameters of the phase were calculated from X-ray diffraction data (Table 2). The indexing of the data shows that it crystallises in hexagonal unit cell ($a=17.452\text{\AA}$ and $c=6.463\text{\AA}$). In order to determine the crystal structure, the theoretical X-ray diffraction data were generated by Treor and Lazy-Pulverix analysis. The d_{cal} values computed from data are in good agreement with the experimental interplanar distances. The data along with the assigned $h k l$ values are given in the Table 2.

Table 2: Powder X-ray Diffraction Data of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$

h	k	l	d_{obs} (Å)	d_{cal} (Å)	I_{obs}
0	0	1	6.463	6.468	100.0
4	0	0	3.778	3.781	1.2
0	0	2	3.232	3.234	8.4
5	0	0	3.014	3.025	3.9
4	2	0	2.849	2.858	17.8
4	2	1	2.607	2.614	3.1
0	0	3	2.154	2.156	7.0
1	0	3	2.134	2.134	0.8
5	1	2	2.080	2.080	0.7
3	1	3	1.914	1.917	19.2
7	1	1	1.910	1.913	11.6
7	2	1	1.775	1.776	0.3
4	2	3	1.718	1.721	1.5
9	0	0	1.679	1.680	0.2
7	3	1	1.645	1.645	8.4
6	3	2	1.641	1.641	6.3
0	0	4	1.616	1.617	13.2
3	2	4	1.464	1.465	0.6
5	0	4	1.424	1.426	1.5
7	2	3	1.402	1.403	2.9

$$a = b = 17.452\text{\AA}$$

$$c = 6.463\text{\AA}$$

$$\alpha = \beta = 90^\circ$$

$$\gamma = 120^\circ$$

Magnetic susceptibility studies

The molar magnetic susceptibility measurements as a function of temperature suggest that the phase is diamagnetic and magnetic susceptibility is temperature independent.

Electric Transport Properties

The electrical resistivity of the phase ($\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$), as function of temperature, is plotted ($\log \rho$ versus $1/T$) in figure 2. The negative temperature co-efficient of resistivity and the values of the specific resistance (Table.3) suggest that the phase is semi-conductor in nature and the electrical conduction occurs via thermal activated mechanism. However, as a result of change in slope at around 344 K the energy of activation (E_a) shows variation.

Table 3: Specific resistance ($\log \rho$) of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$ as function of temperature (K).

Temperature (K)	Specific resistance ρ (ohm cm)
309	1.276×10^5
314	9.437×10^4
319	7.009×10^4
324	5.257×10^4
329	4.081×10^4
334	3.233×10^4
339	2.454×10^4
344	1.926×10^4
349	1.521×10^4
354	1.221×10^4
359	1.013×10^4
364	8.00×10^3
368	6.759×10^3
373	5.632×10^3
378	4.590×10^3
383	3.980×10^3
387	3.304×10^3
392	2.696×10^3
397	2.102×10^3
402	1.834×10^3
407	1.524×10^3
411	1.368×10^3
416	1.156×10^3
421	1.021×10^3
426	9.168×10^2
431	8.453×10^2
436	7.785×10^2
441	7.246×10^2
446	7.043×10^2
451	6.459×10^2
456	6.008×10^2
460	5.182×10^2

Table4: Magnetic and Electric Transport Parameters of ($\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$) phase.

Phase	μ_{eff} (B.M)	μ_{theo} (B.M)	E_a (eV)
$\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$	Diamagnetic	-	0.50 (309-421 K) 0.30 (426-460K)

Thermal Analysis:

The thermogravimetric analysis (TGA) of $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$ fig. 3 suggests that the phase is almost stable between 403K and 673K and above 673K and above 673 K there is mass loss of about 4%. In the temperature range 489.4K – 498.8K there is an endothermic peak with peak temperature 496.9 K in the differential analysis (DTA) curve. The peak is attributed to phase transformation process which in this case is considered to be polymerization process.

Conclusion

A new phase with the composition $\text{Mo}_{0.5}\text{Si}_{0.5}\text{Se}_{1.94}$ has been synthesised by the standard ceramic method. On the basis of Lazy-Pulverix analysis of the X-ray diffraction data it is concluded that the phase crystallises in hexagonal unit cell. The molar magnetic susceptibility measurements as a function of temperature suggest that the phase is diamagnetic and magnetic susceptibility is temperature independent. The study of electrical resistivity in the temperature range 300K-500K shows that the compound is an electrical semi-conductor and conduction occurs via thermal activated mechanism, however an anomaly at around 344 K shows a variation in the energy of activation (E_a). The thermal analysis suggests that there is mass loss of about 4% above 673 K and at low temperature (496.9K), there is phase transformation which results polymerisation process.

Acknowledgements

Thanks are due to the UGC, New Delhi for financial support, University of Delhi for thermal analysis, IIT Bombay for XRD studies and Department of Chemistry, University of Jammu, Jammu for providing requisite facilities.

References:

- [1] Folmer J.C.W, Jellinek F. and Calis G.H.M. (1988).The Electronic Structure of Pyrites, Particularly CuS_2 and $\text{Fe}_{1-x}\text{Cu}_x\text{Se}_2$: An XPS And Mössbauer Study, J. Solid State Chem., 72, 137-144.
- [2] Li Jing, Guo H.Y., Proserpio D.M. and Sironi A. (1995).Exploring Tellurides: Synthesis and Characterization of New Binary, Ternary, and Quaternary Compounds, J. Solid State Chem., 117, 247-255.
- [3] Fischer P. Plambeck, Abriel W. and Urand W. (1989). Preparation and Crystal Structure of $\text{RESe}_{1.9}$ (RE = Ce, Pr), J. Solid State Chem., 78, 164.
- [4] Pardo M.P., Gardette M.F., Dung N. and Flahaut J. (1991). Insertion d'oxygène dans les composés de type $\text{Cu}_{0.5}\text{RTe}_2$ et $\text{Cu}_{0.5}\text{RTe}_{1.75}$ (R = La et Nd), J. Solid State Chem., 94,121-129.
- [5] Huan G. and Greenblatt M. (1987). New $\text{A}_x\text{Nb}_6\text{Se}_8$ (A= Na, K, Rb, Cu, Ag, Zn, Cd, Pb) phases with the Nb_3Te_4 structure, Mater. Res. Bull., 22, 505.



- [6] Huan G. and Greenblatt M, (1991). Inorganic Syntheses, Non molecular Solids, Mater. Res. Bull., 22, 943.
- [7] Limatta E.W. and Ibers J.A. (1987). Synthesis, Structure, and Physical Properties of the New Layered Ternary Chalcogenide NbNiTe₅, J. Solid State Chem., 71, 384.
- [8] Elwell W.T. and Gidley J.A.F., Atomic Absorption Spectrophotometry, Pergamon, 102. (1961).
- [9] Robinson J.B., Atomic Absorption Spectroscopy, Marcell Dekker, New York. (1966).
- [10] Khopkar S.M., Basic Concept of Analytical Chemistry, Wiley Eastern Limited. (1985).
- [11] Taguchi H., Nagao M. and Shimada M. (1988). Metal-Insulator Transition in the System (Nd_{1-x}Cax) MnO_{2.99} (0.5 ≤ x ≤ 0.9), J. Solid State Chem., 76, 284-289.
- [12] Jeffery G.H., Vogel's Text Book of Quantitative Chemical Analysis (ELBS, Longman, Essex). (1989).
- [13] Yvon K., Jeitschko W. and Perthe E., Lazy Pulverix, Laboratory De Crystallographic Aux Rayons-X, Universities De Genève, Genève, Switzerland. (1977).
- [14] Werner P. E., Treor - 4 Deptt. of Structural Chemistry, Arrhenius Laboratory University of Stockholm, Sweden. (1984).
- [15] Noel H., Potel M., Troc R. and Shlyk. L. (1996). Crystal Structure and Physical Properties of β USe₂ and USe_{2-x}Tex (x= 0.24 and 0.72), J. Solid State Chem., 126, 22-26.
- [16] Selwood P.W., Magnetochemistry, (Interscience, New York), (1943).
- [17] Goodenough J.B. (1963). Magnetism and the Chemical Bond (Interscience, New York).
- [18] Resistivity of Semiconductors by Four Probe Method at Different Temperature Scientific Equipment and Services Roorkee 247667 U.P).
- [19] Wieder H.H and Elsevier Amsterdam., Laboratory Notes on Electrical and Galvanomagnetic Measurements. 67, 53-55. (1979).
- [20] J.G. Dum and L.C. Mackey, J Thermal Anal., 37 (1991) 2143.
- [21] El-H.M. Diefallah, Thermochim. Acta 202 (1992) 1.